

~NKFUST--傅兆章

材 料 與 製 造

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二、金屬材料製造系統 ~NKFUST--傅兆章

(一)

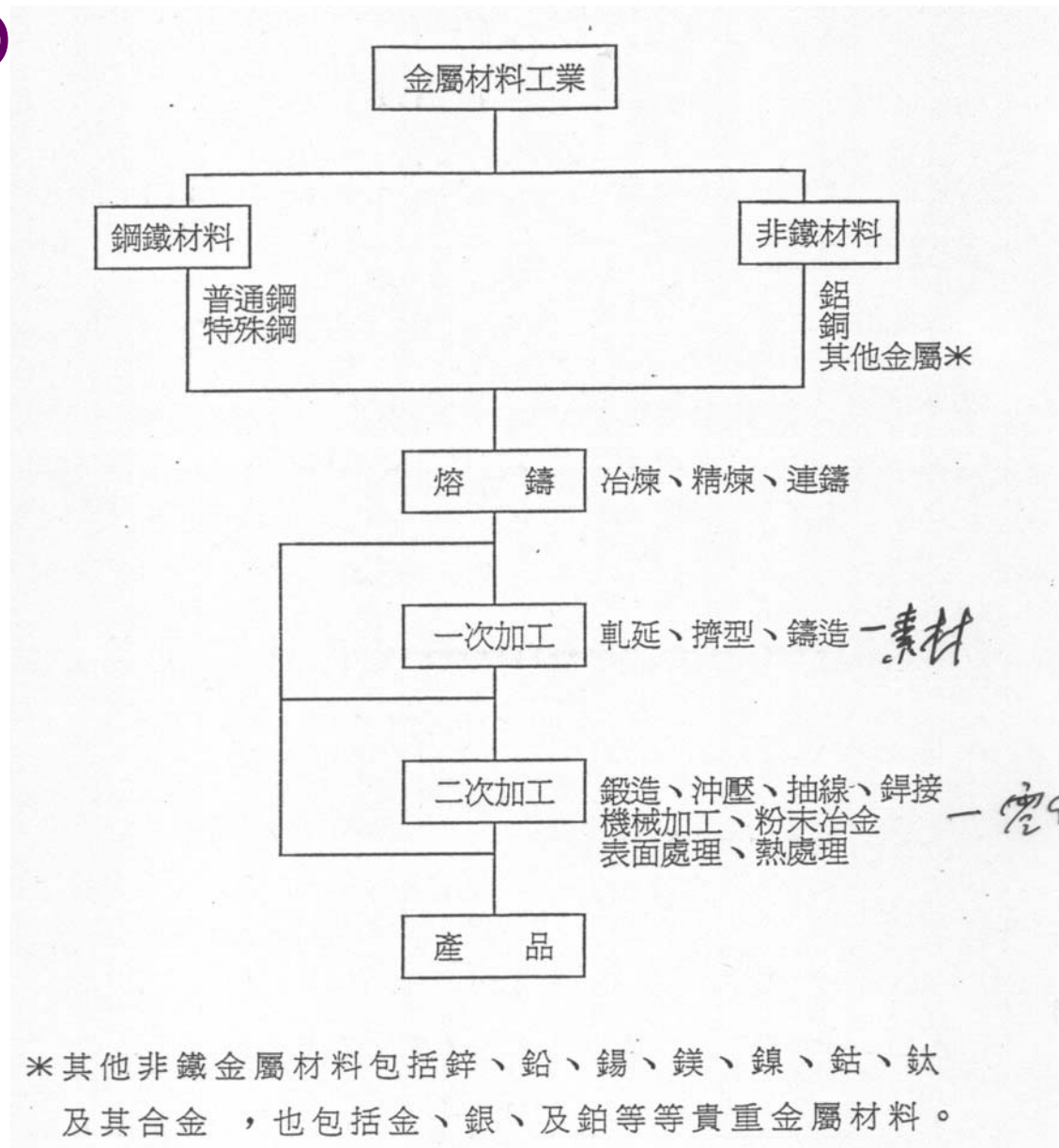


圖1-1 金屬材料工業之產業結構圖

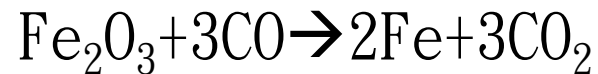
(二)鐵及鋼的製造系統 ~NKFUST--傅兆章

原料：

(1)鐵礦石：磁鐵礦赤鐵礦褐鐵礦65% Fe

(2)焦炭→加熱至1150°C在Quench。因未還原鐵礦，
有副產品。

(3)石灰石→去雜質及造渣，形成slag



煉鐵→高爐1650°C→鐵水(含高碳)造pig iron(4% C+1.5% Si、1% Mn)

煉鋼→轉爐或平爐

B. O. F：將生鐵+廢鋼或添合金元素

吹氧作 $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$ ， O_2 減少『C』含量

速鑄→板材、棒材或型鋼

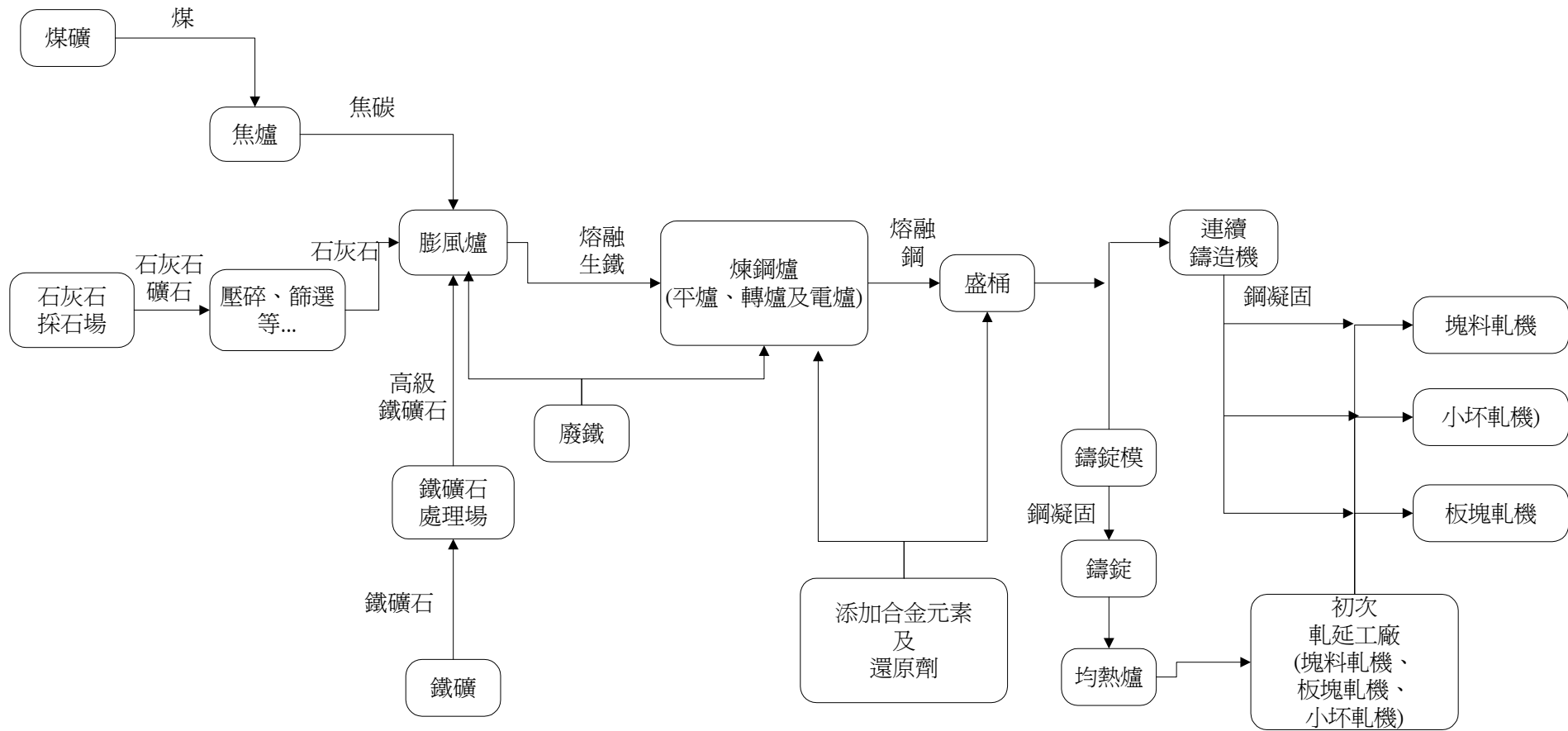


圖9.5 鋼鐵原料轉變成各種鋼鐵產品形式(不包含有塗層之產品)之
主要步驟流程圖

[資料來源：H.E. McGannon (ed.), “The Making Shaping, and Treating of Steel,” 9th ed., United States Steel Corp., 1971, p.2.]

(三) 鋼鐵的連鑄

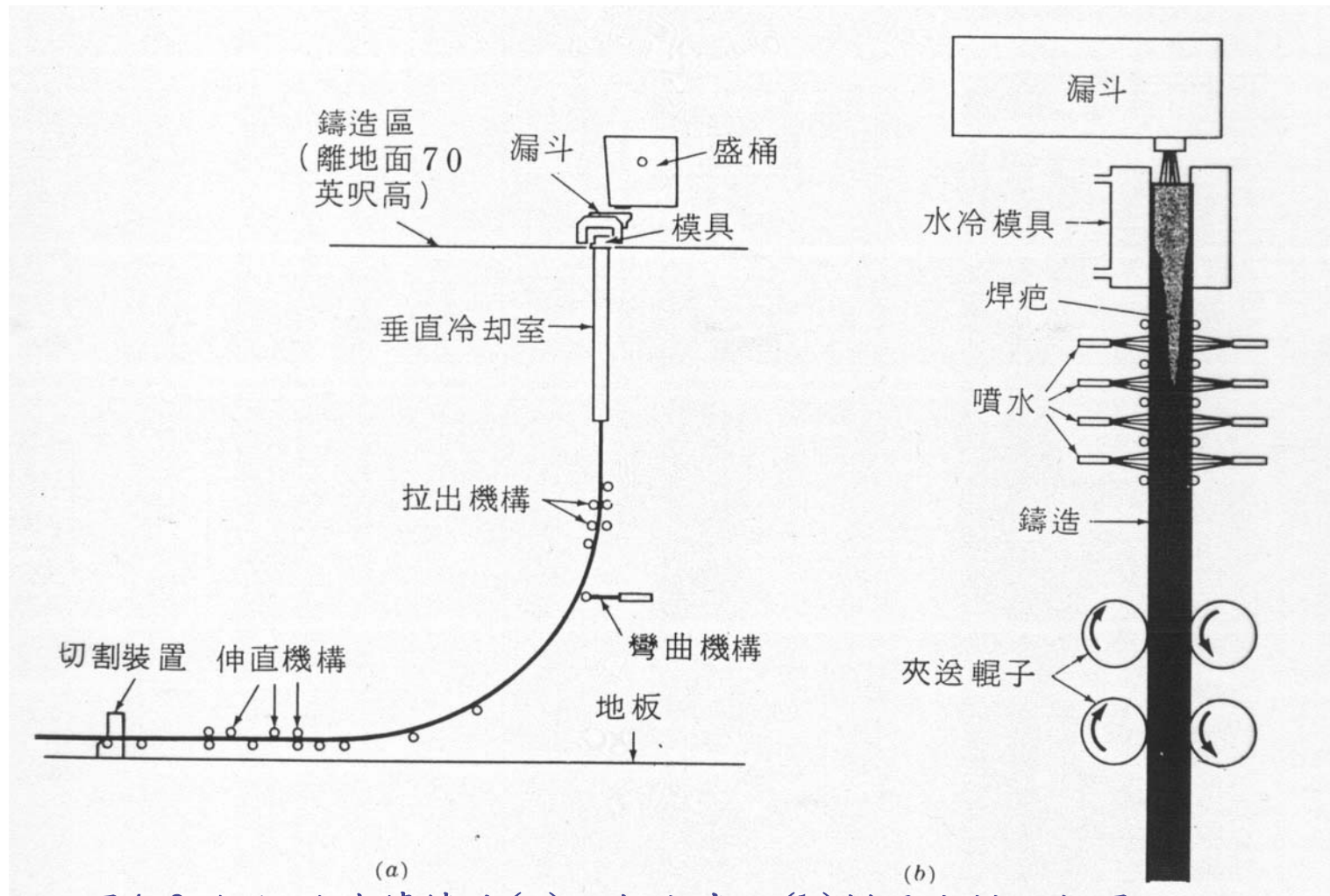
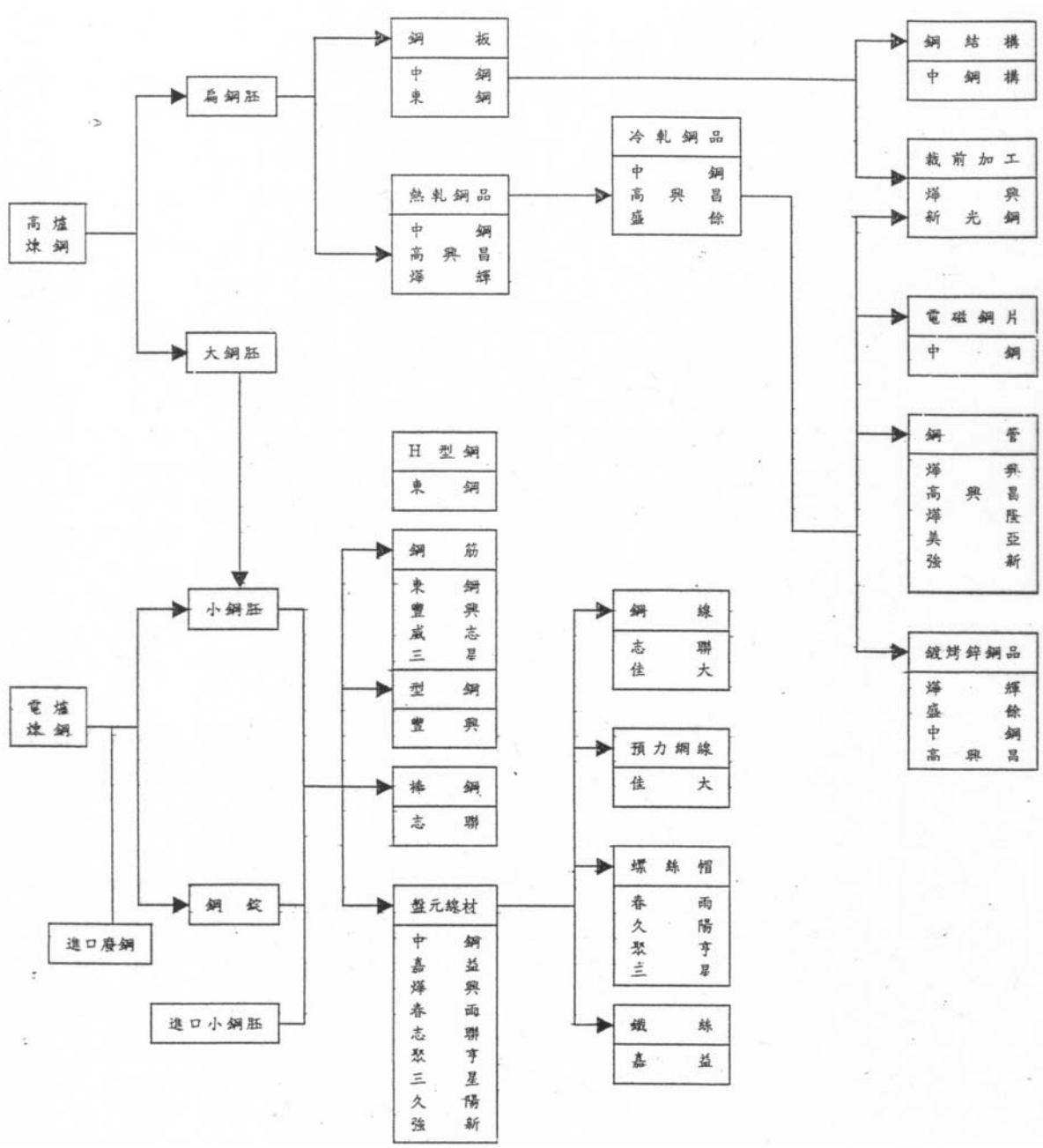


圖4.9 鋼錠的連續鑄造(a)一般方式；(b)模具安排細部圖。

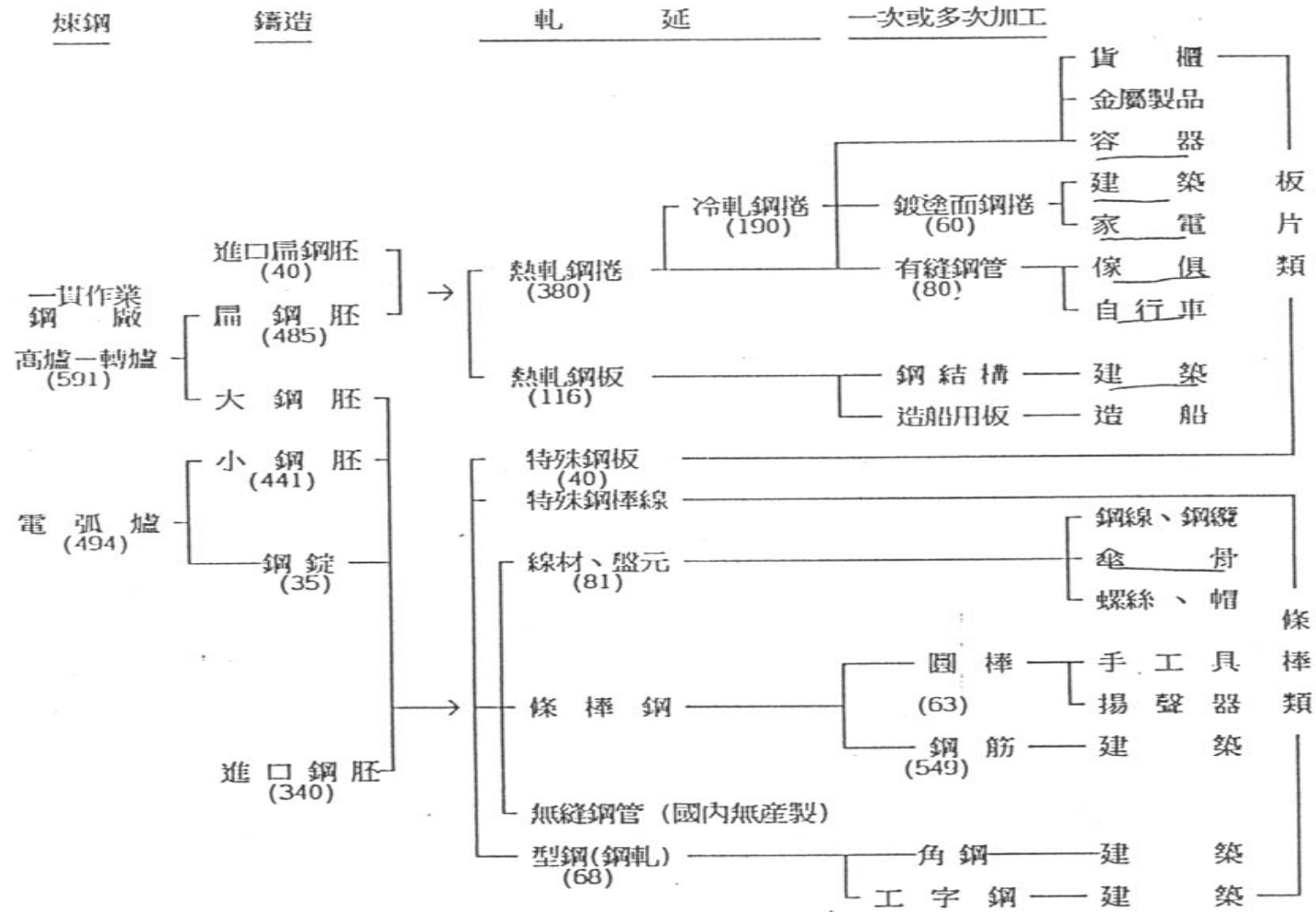
[資料來源：H.E. McGannon (ed.), "The Making Shaping, and Treating of Steel," 9th ed., United States Steel Corp., 1971, pp.707~708.]

(四) 碳鋼產業關聯圖 ~NKFUST--傅兆章



(五)我國鋼鐵產業

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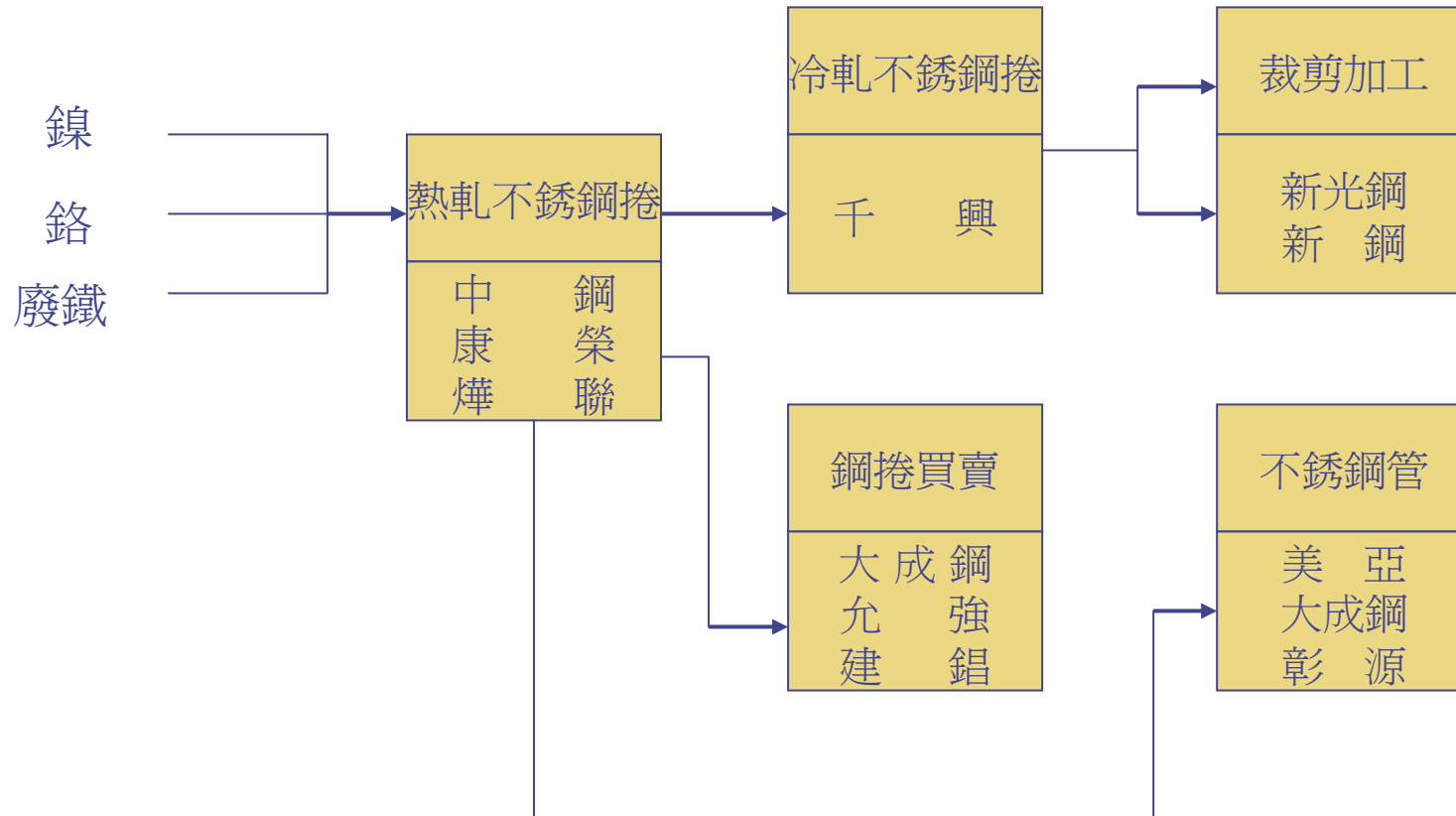


註：1.括號內數字為民國八十年國內產量(萬公噸)
 2.我國一貫作業鋼廠僅中鋼公司 1家；電弧爐煉鋼廠約 43家；單軋鋼廠約 221家

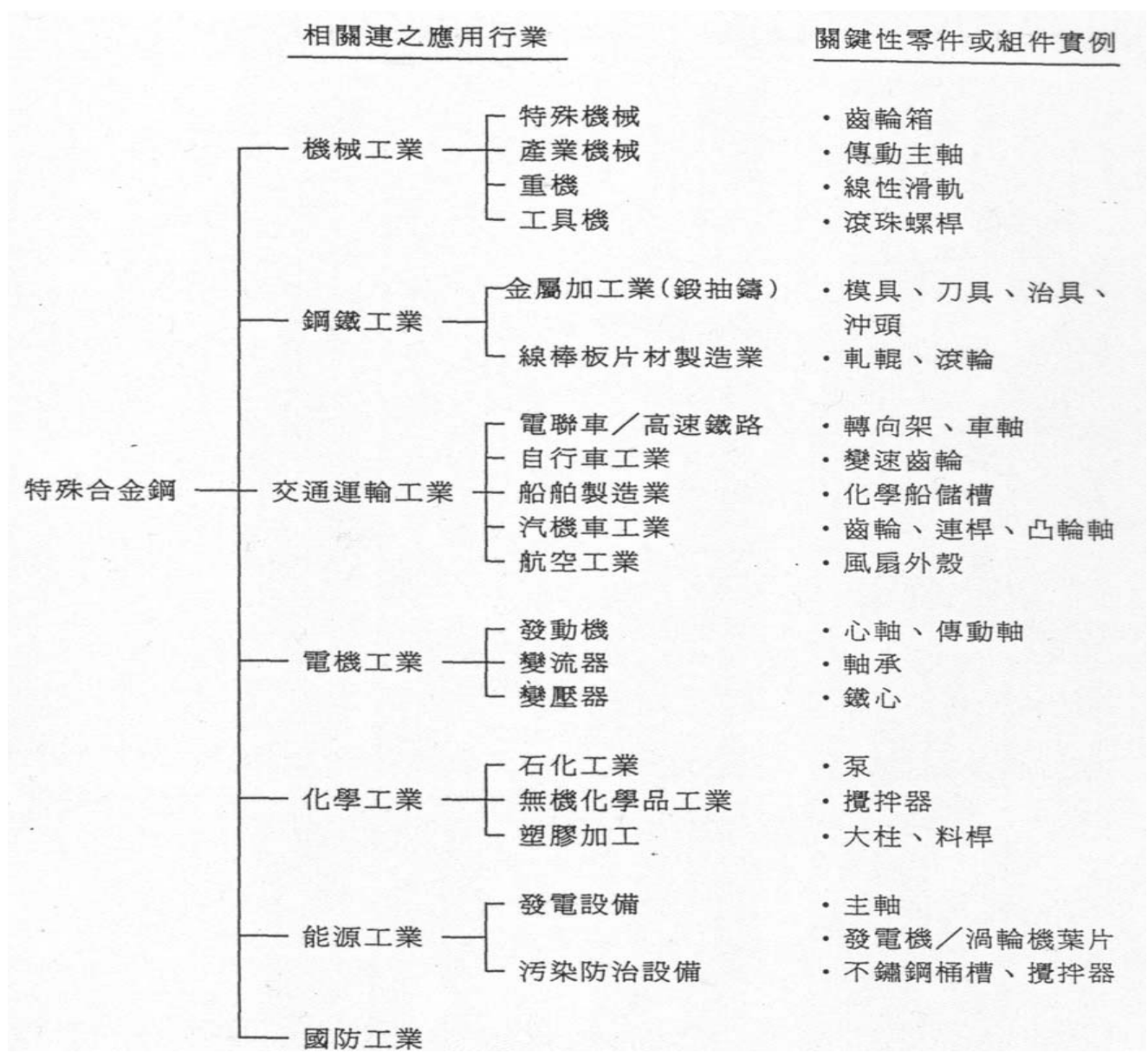
資料來源：中國鋼鐵公司與鋼鐵公會提供

圖3-1 我國鋼鐵產業結構圖

(七)不鏽鋼產業關連圖

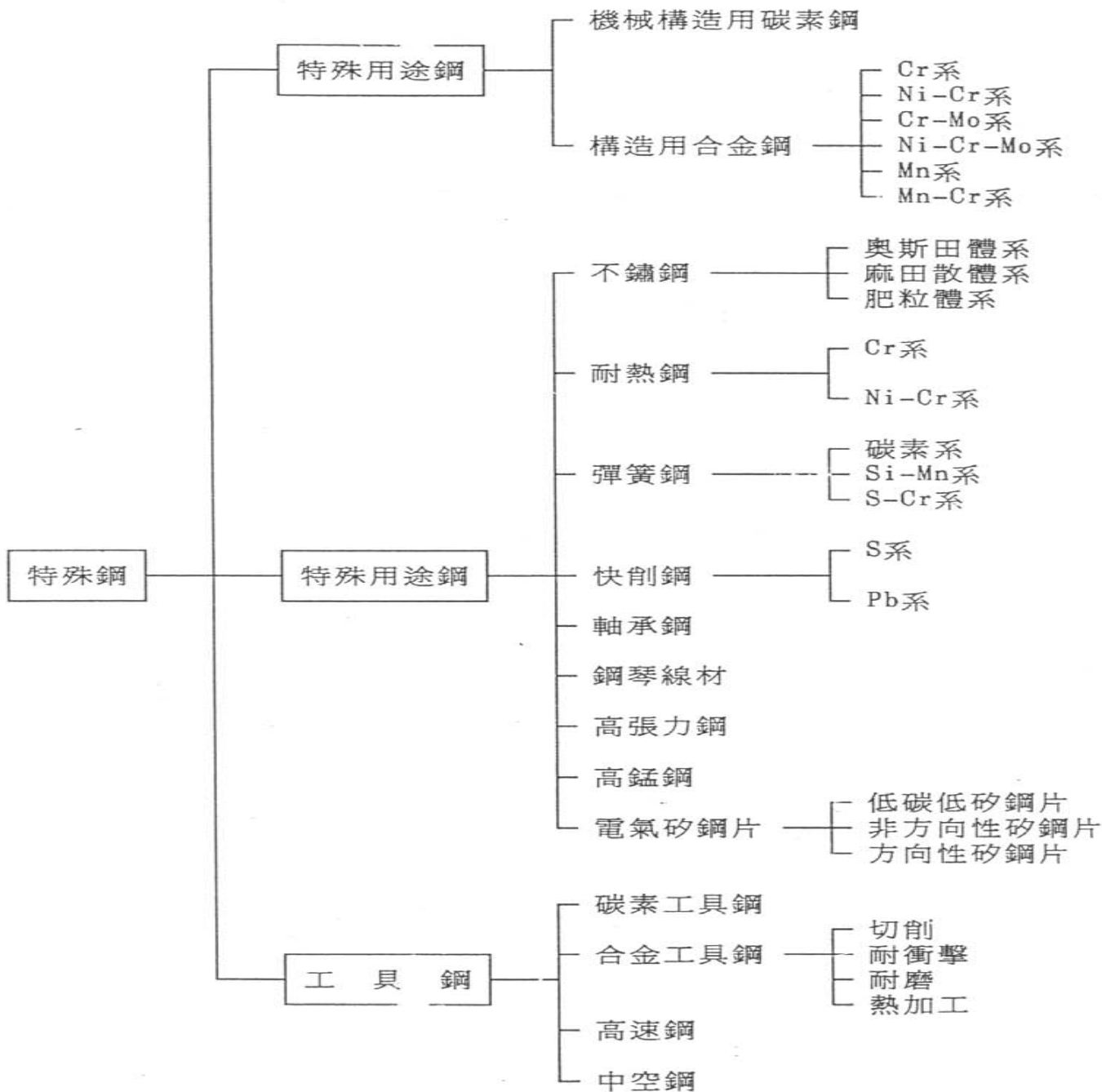


(八)特殊合金鋼產業關連圖



(九)特殊鋼的分類

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(十) Fe-Fe₃C 平衡圖

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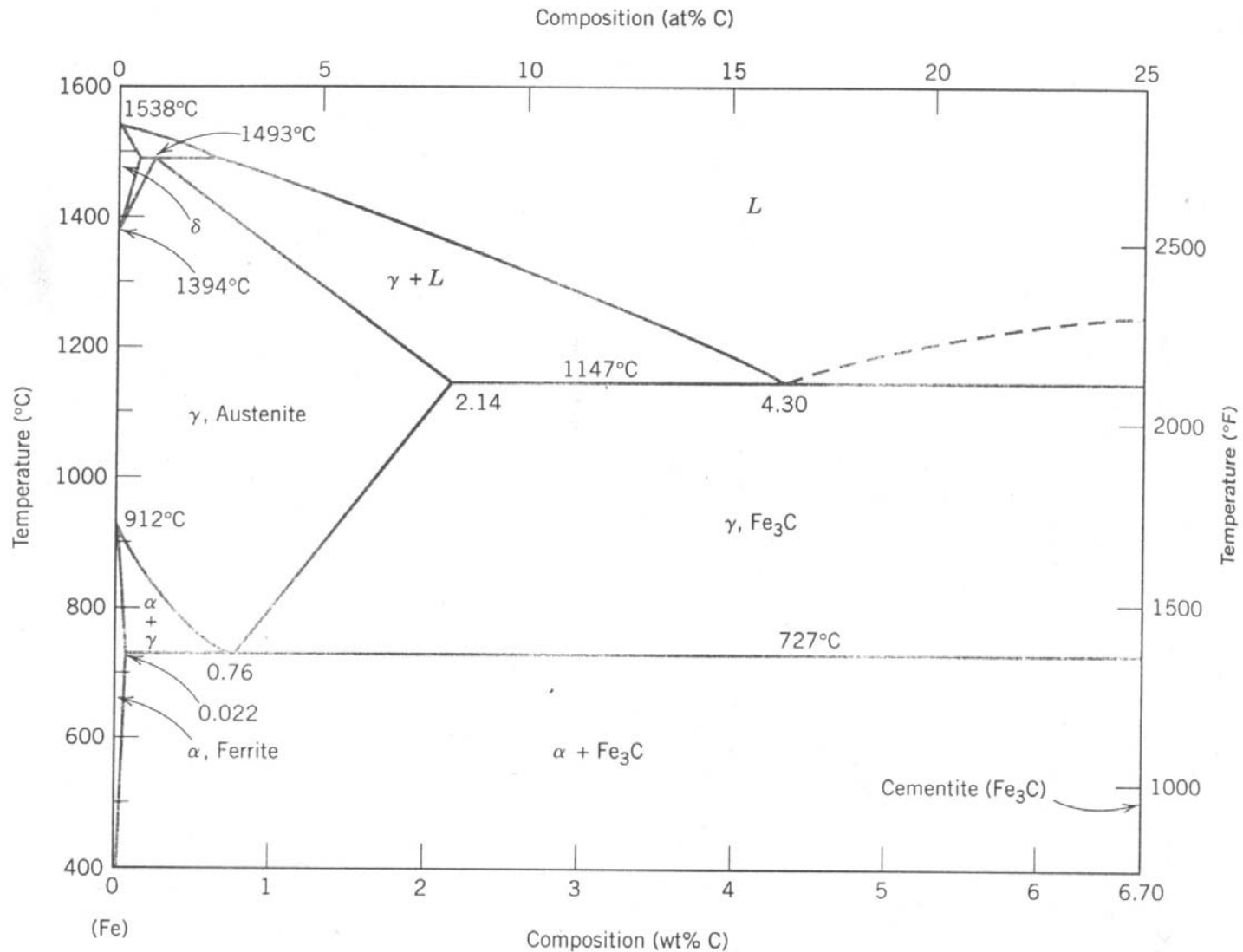


Figure 9.22 The iron-iron carbide phase diagram.(Adapted from Binary Alloy Phase Diagrams, 2nd edition, Vol. 1, T. B. Massalski, Editor-in-Chief, 1990. Reprinted by permission of ASM International, Materials Park, OH 44073-0002.)

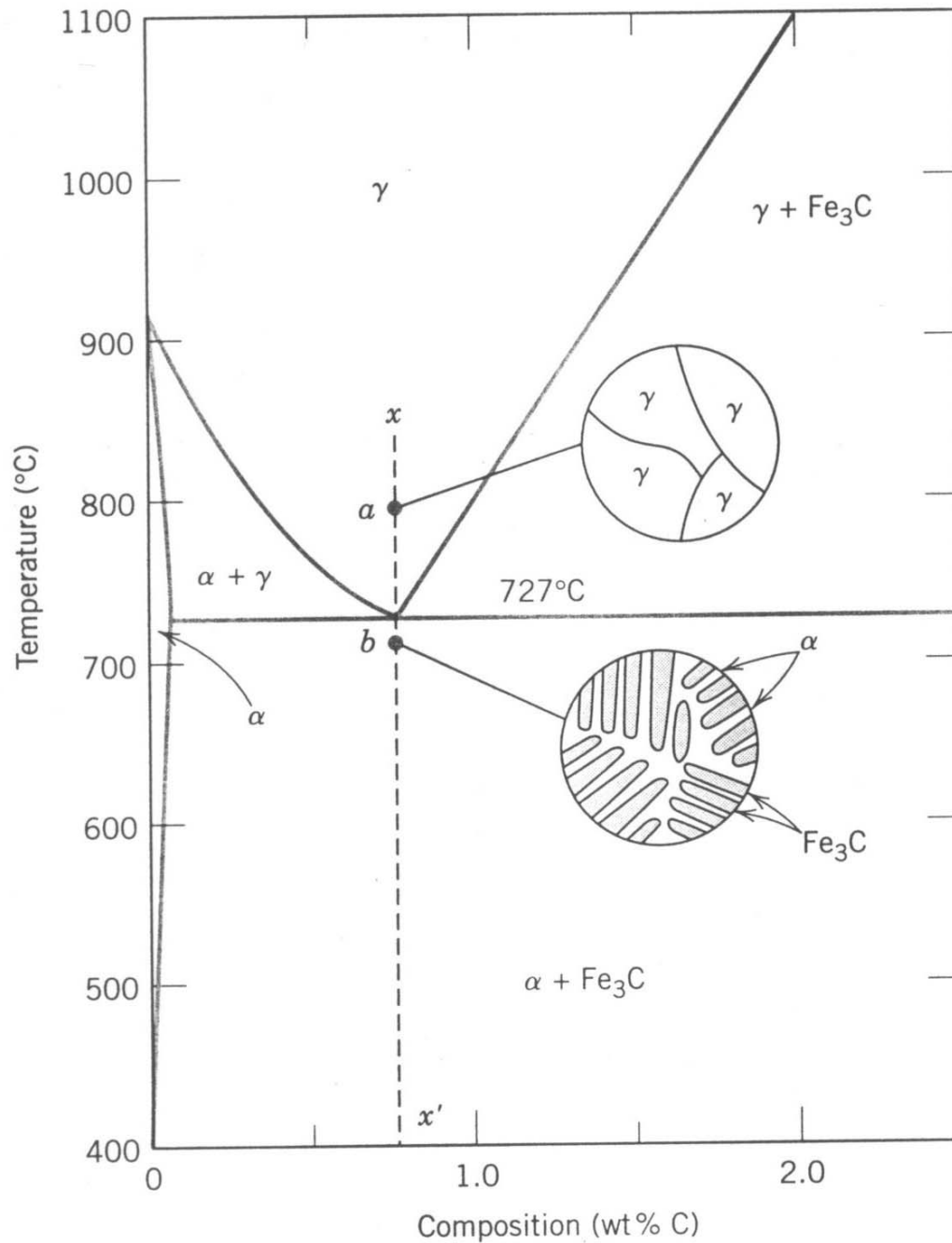


Figure 9.24 Schematic representations of the microstructures for an iron-carbon alloy of eutectoid composition(0.76wt% C) above and below the eutectoid temperature



Figure 9.25
Photomicrograph of a eutectoid steel showing the pearlite microstructure consisting of alternating layers of α ferrite (the light phase) and Fe_3C (thin layers most of which appear dark). 500 \times . (Reproduced with permission from Metals Handbook, Vol. 9, 9th edition, Metallography and Microstructures, American Society for Metals, Materials Park, OH, 1985)

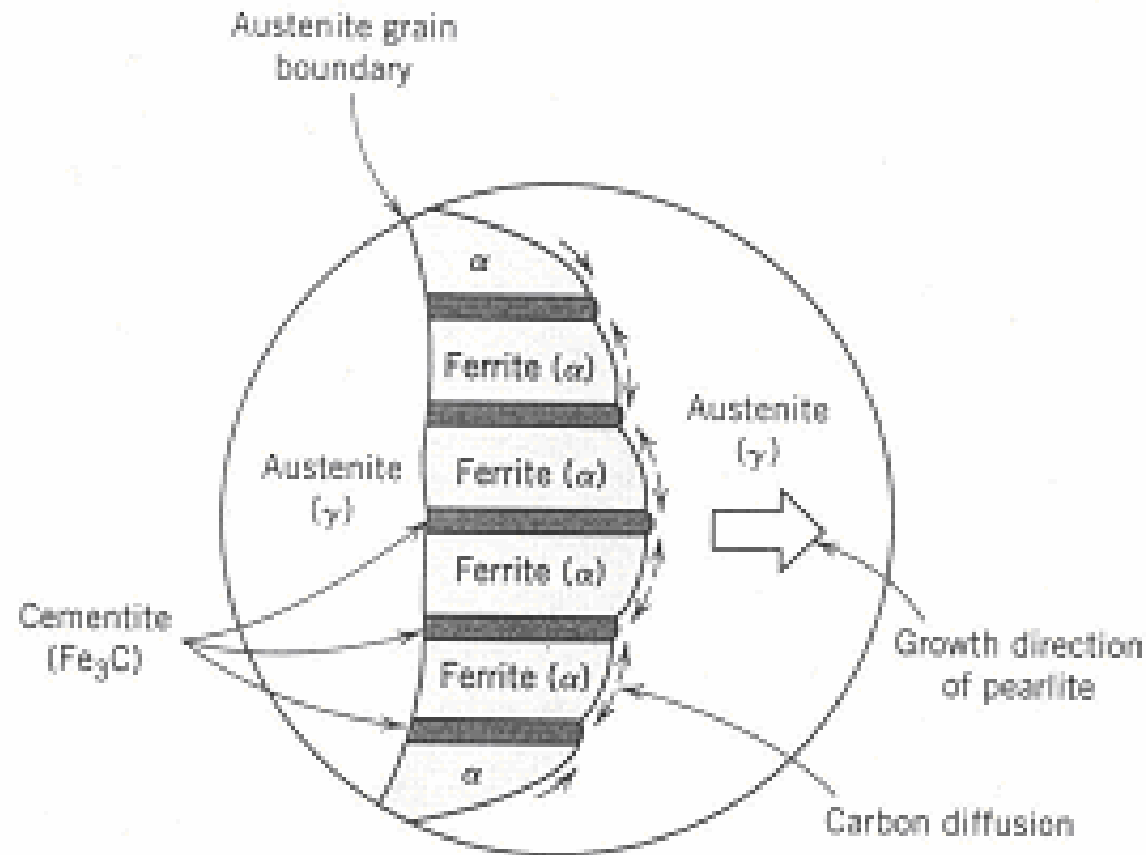


Figure 9.26 Schematic representation of the formation of pearlite from austenite; direction of carbon diffusion indicated by arrows

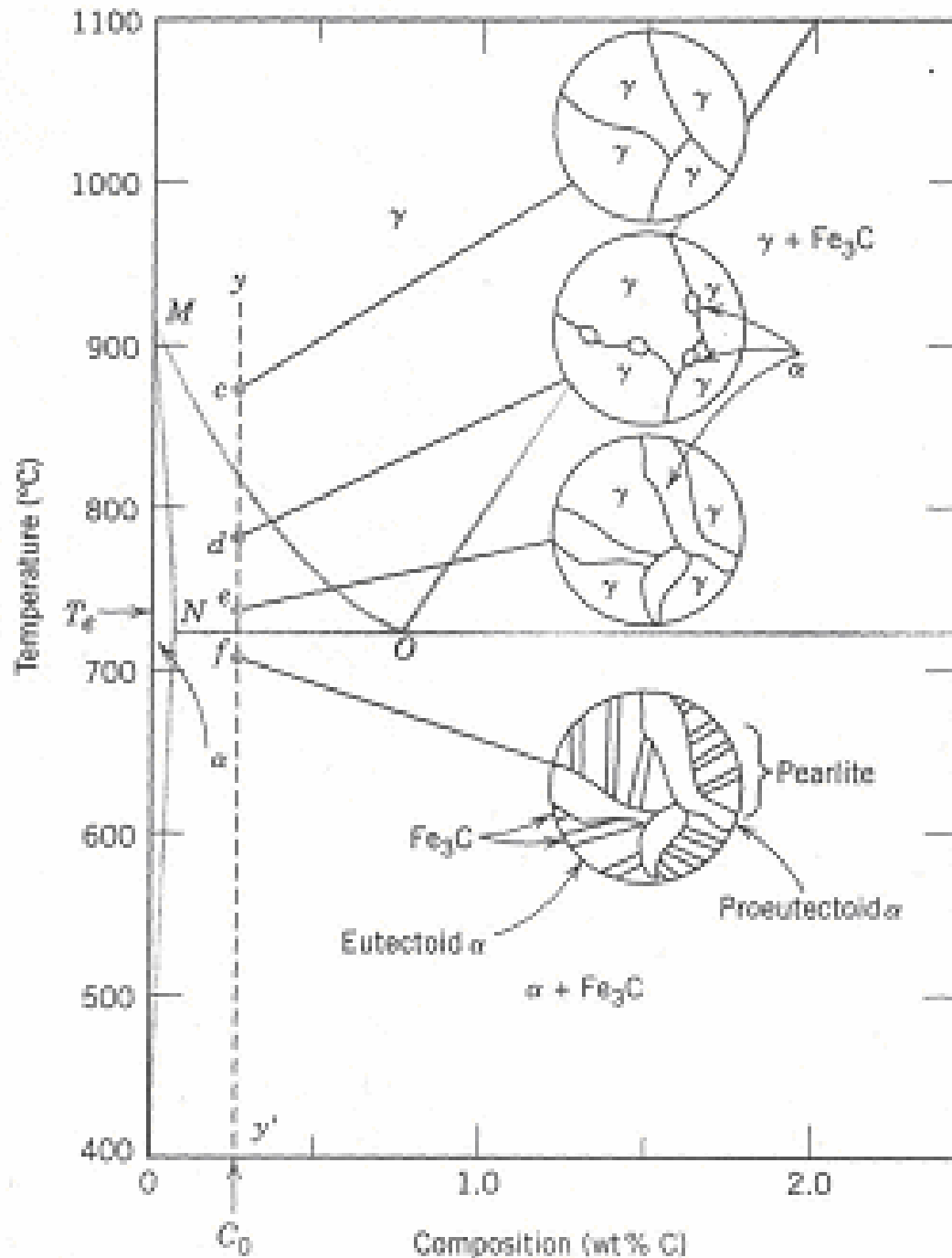


Figure 9.27 Schematic representation of the microstructures for an iron-carbon alloy of hypoeutectoid composition C_0 (containing less than 0.76wt% C) as it is cooled from within the austenite phase region to below the eutectoid temperature

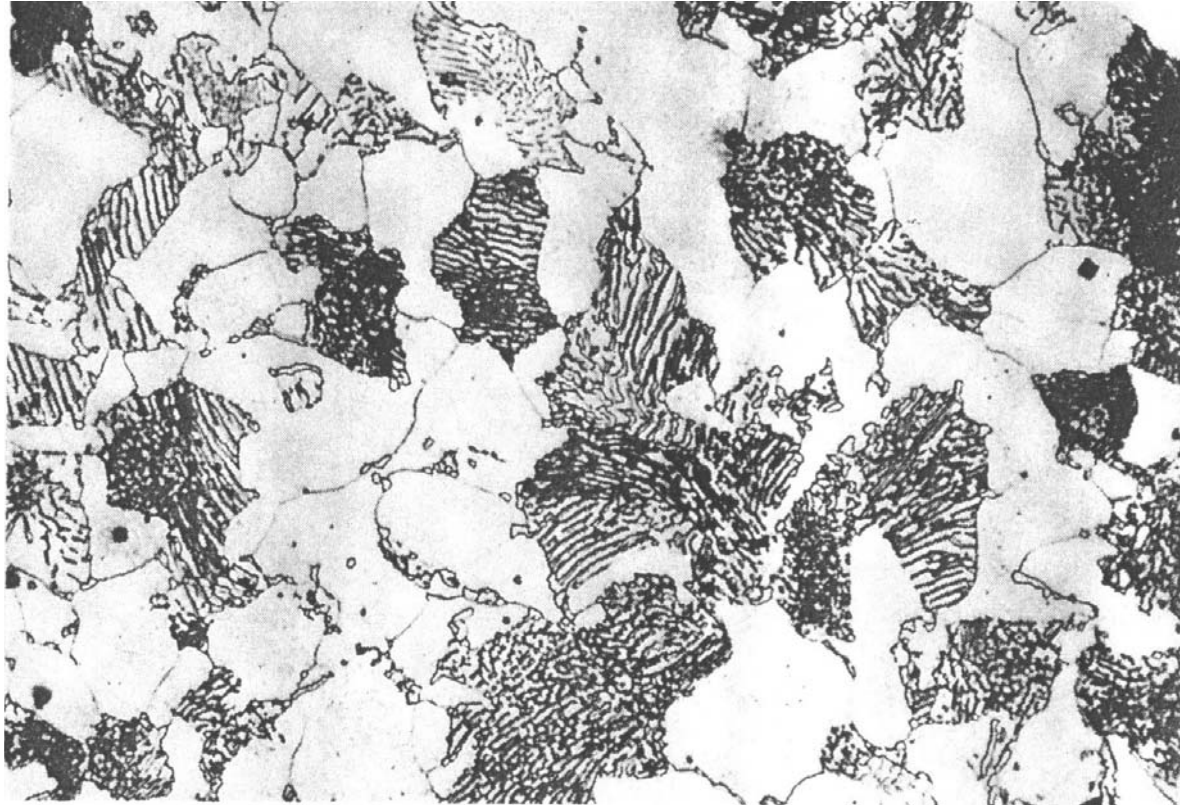


Figure 9.28 Photomicrograph of a 0.38wt% C steel having a microstructure consisting of pearlite and proeutectoid ferrite. 635 \times (Photomicrograph courtesy of Republic Steel Corporation)

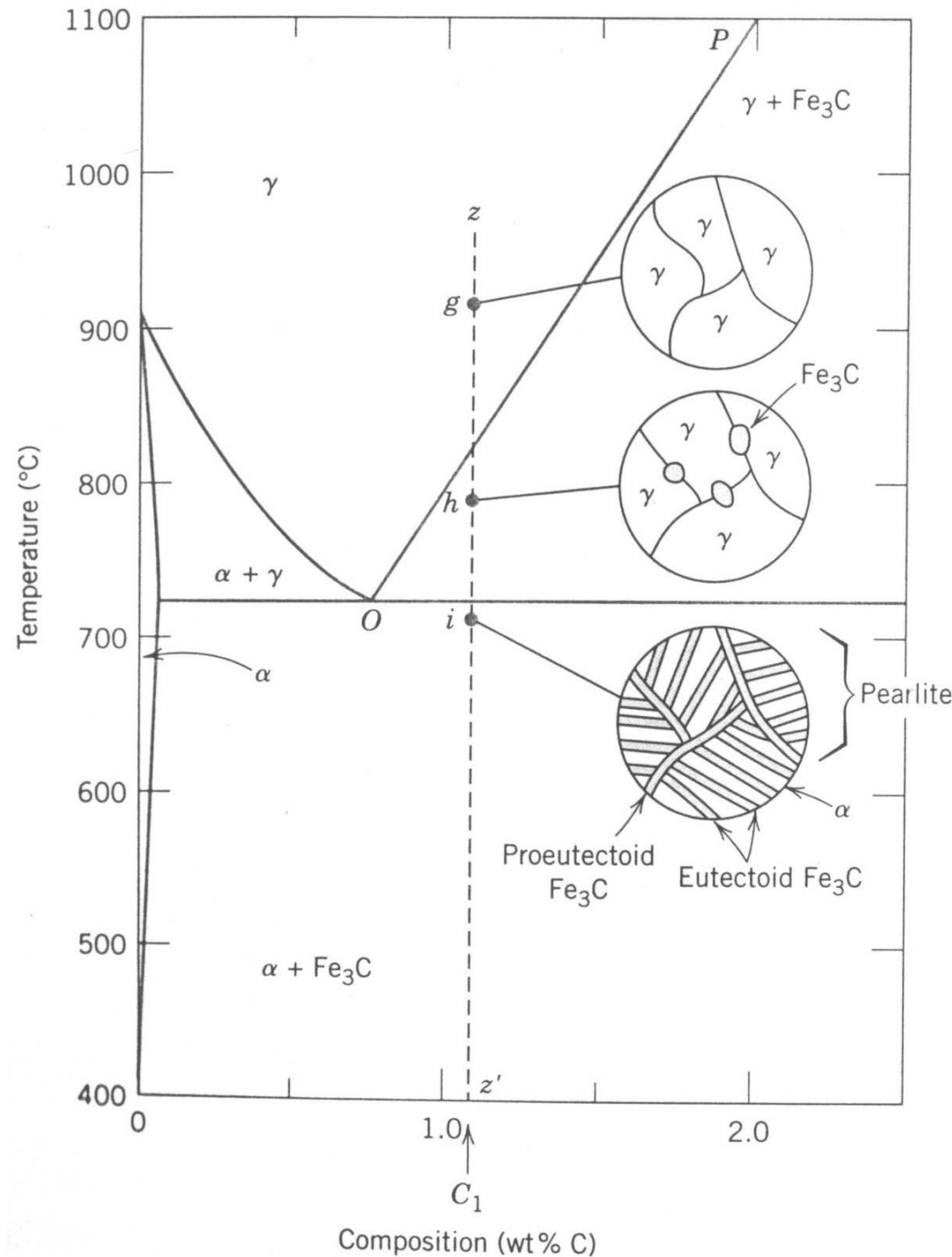


Figure 9.30 Schematic representations of the microstructures for an iron-carbon alloy of hypereutectoid composition C_1 (containing between 0.76 and 2.14wt% C), as it is cooled from within the austenite phase region to below the eutectoid temperature



Figure 9.31 Photomicrograph of a 1.4wt% C steel having a microstructure consisting of a white proeutectoid cementite network surrounding the pearlite colonies. 1000 \times . (Copyright 1971 by United States Steel Corporation.)

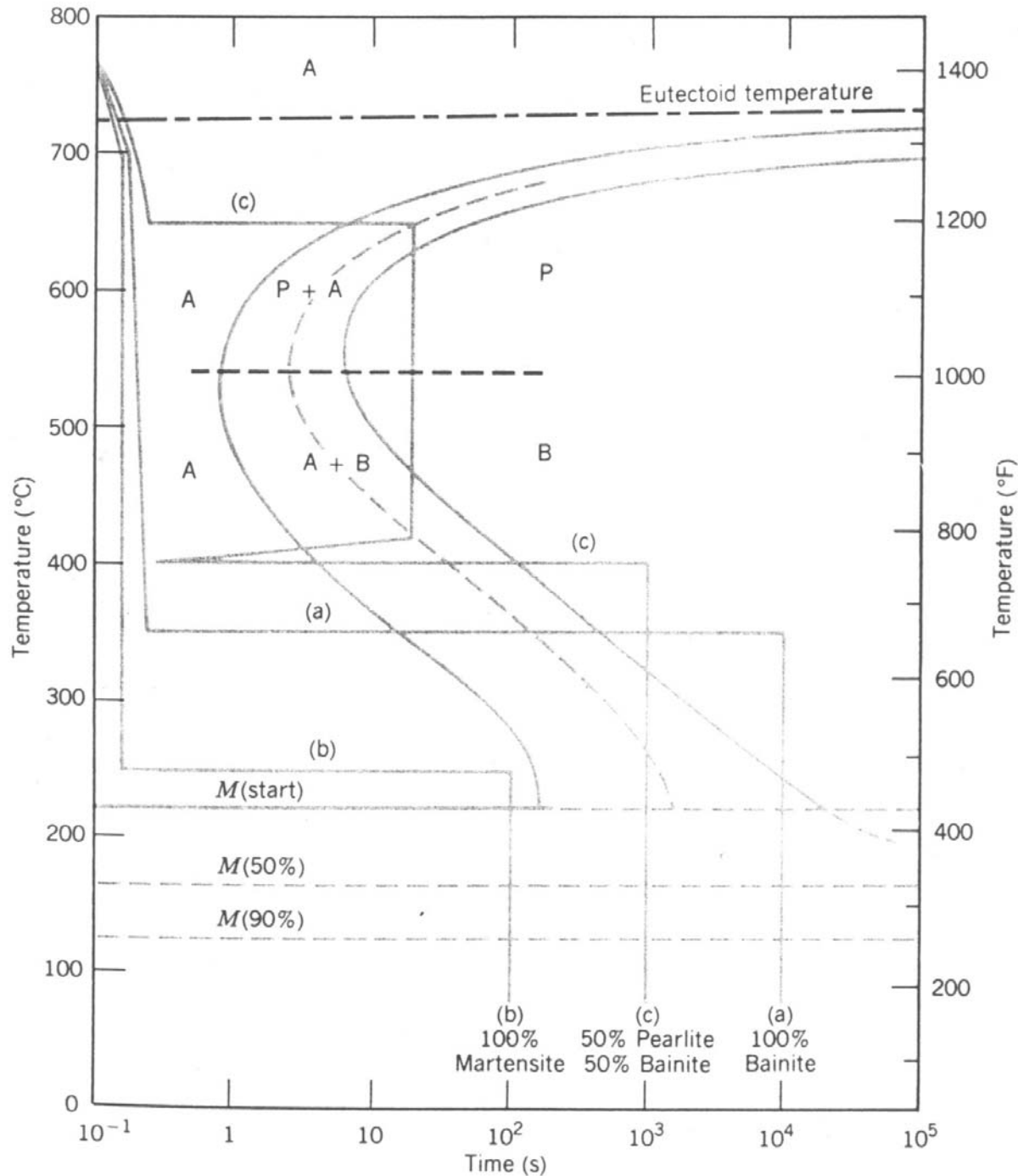


Figure 10.15 Isothermal transformation diagram for an iron-carbon alloy of eutectoid composition and the isothermal heat treatments (a), (b), and (c) in Example Problem 10.1.

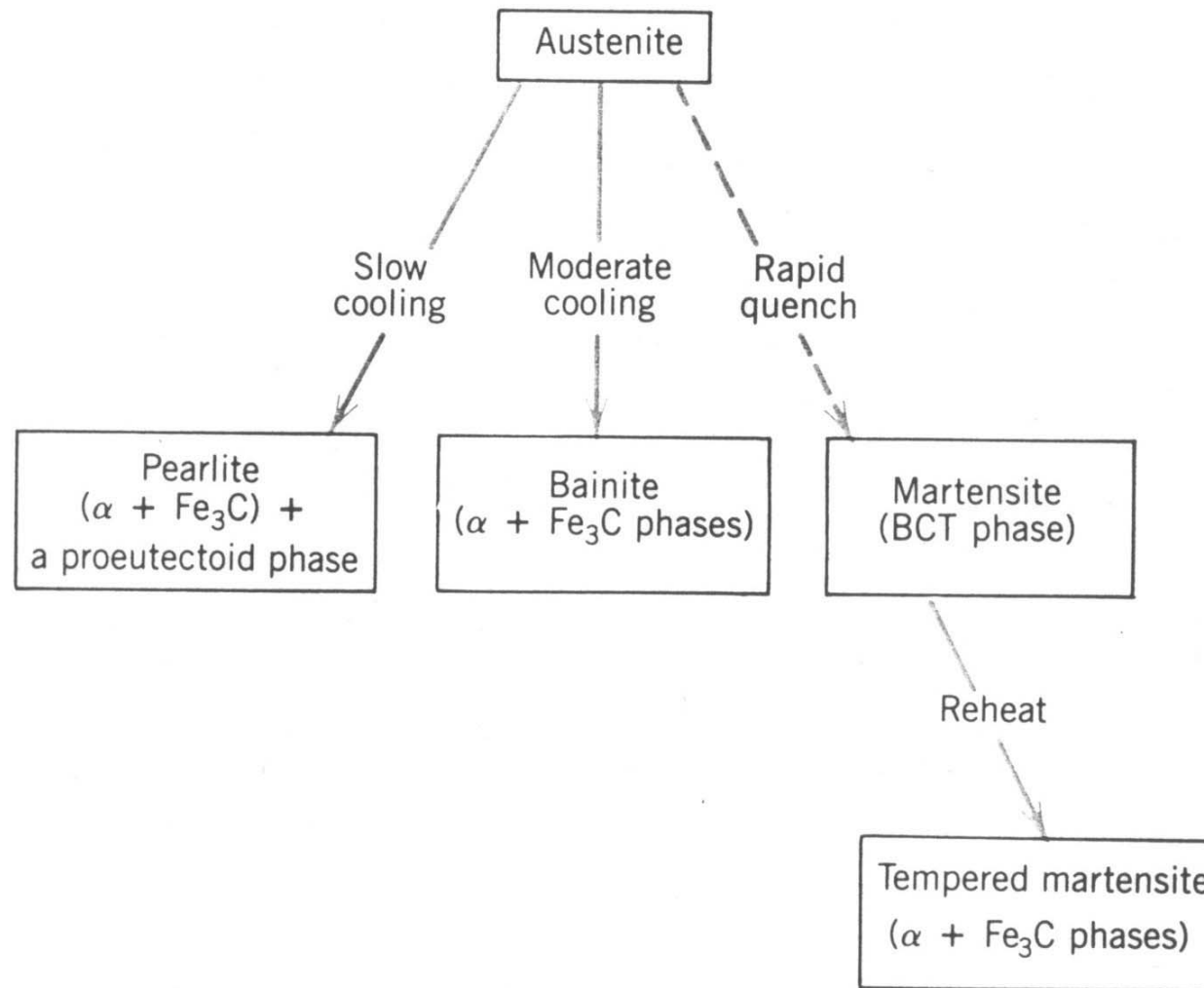


Figure 10.26 Possible transformations involving the decomposition of austenite. Solid arrows, transformations involving diffusion; dashed arrow, diffusionless transformation.

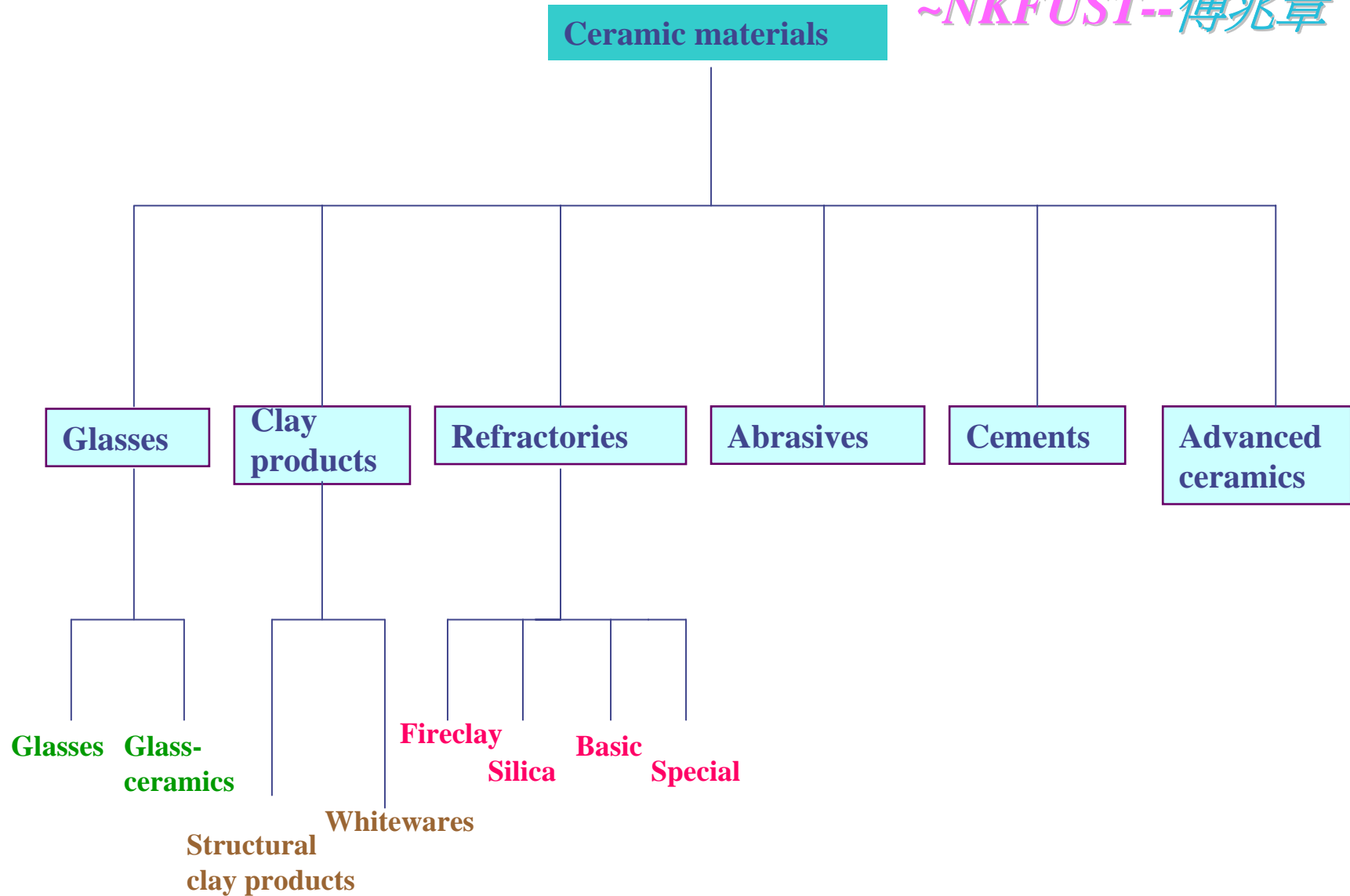


Figure 14.1 Classification of ceramic materials on the basis of application

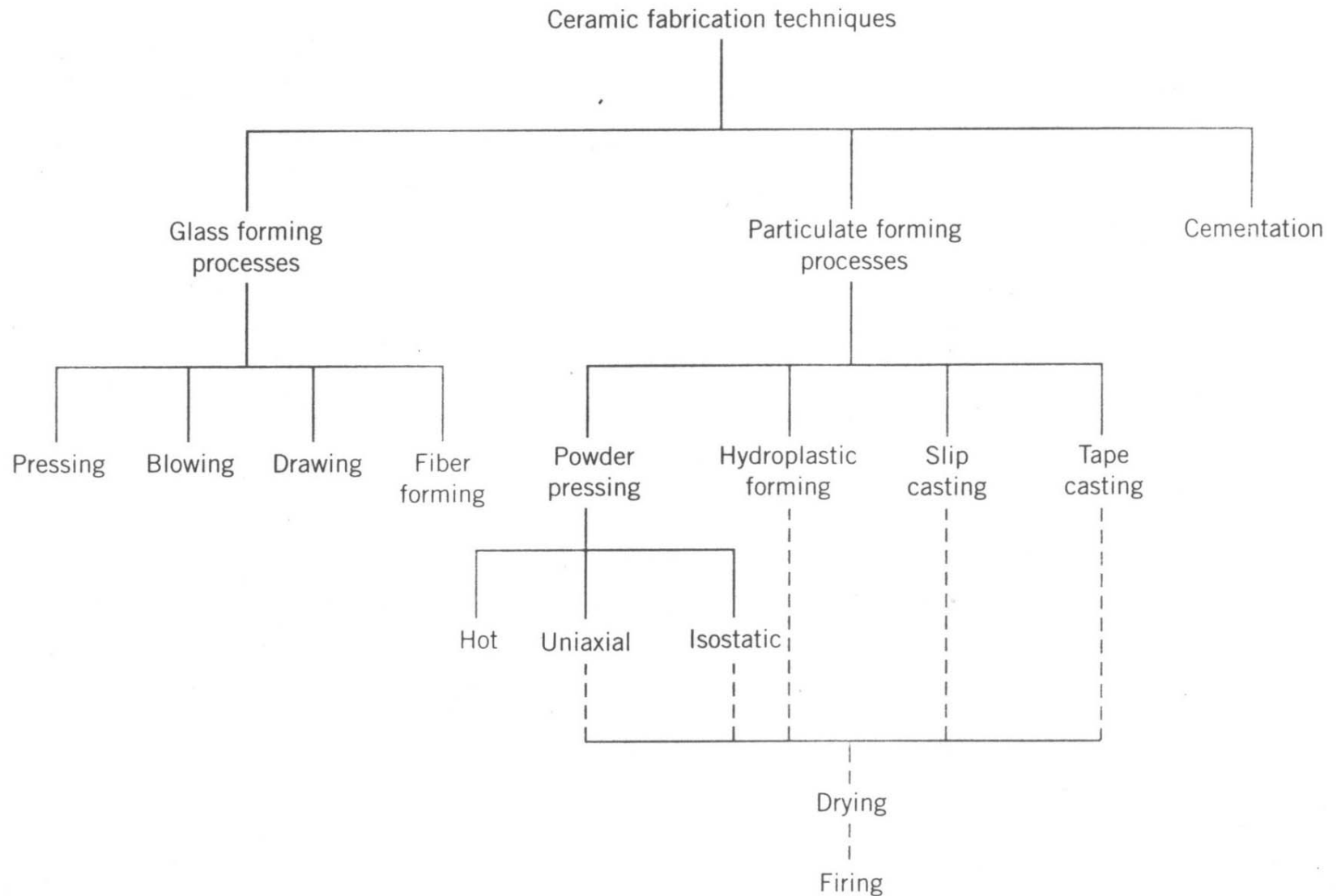


Figure 14.2 A classification scheme for the ceramic-forming techniques discussed in this chapter

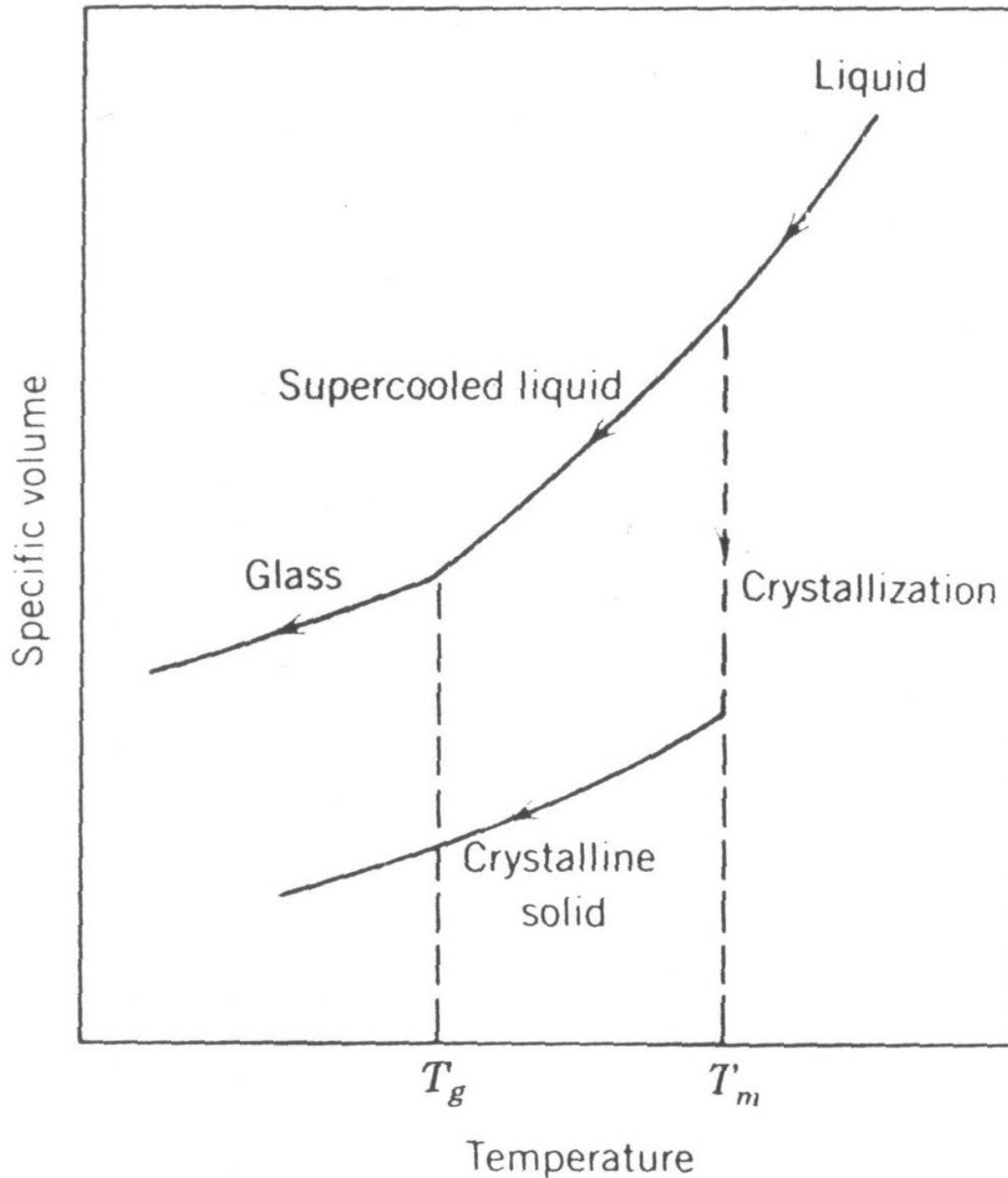


Figure 14.3 Contrast of specific volume-versus-temperature behavior of crystalline and noncrystalline materials. Crystalline materials solidify at the melting temperature T_m . Characteristic of the noncrystalline state is the glass transition temperature T_g